

# Chapter 11 The Mole

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the mole chapter 11 vocab. mole. avogadro's number. molar mass. percent composition. si base unit used to measure the amount of a substance.  $6.022 \times 10^{23}$ .

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mass in grams of one mole of any pure substance. (g/mol) equal to the atomic mass.

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3. an empirical formula mass is found to be 17 g/mol. the molar mass is found to be 34 g/mol. what is the number of empirical formula units in this molar mass?

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